Arsenic Hydride Radicals

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 $^{3}\Pi_{9}-^{3}\Sigma^{-}$

FLASH photolysis of arsine at ~ 20 mm. Hg pressure has yielded two new groups of sharp absorption bands which are attributed to the AsH and AsH₂ free-radicals. These assignments have been confirmed by the isotope shifts observed using fully deuterated arsine, AsD₃.

Arsine was prepared by the method described by Jolly,¹ and the deuteroarsine by the action of deuterium oxide on aluminium arsenide. The infrared spectra of the AsH_3 and AsD_3 showed negligible absorption from impurities. The electronic spectra were photographed in the first and second orders of a 21 ft. concave grating spectrograph with an absorption path of 1 metre. The

half-lives of both species under the above conditions are $\sim 60 \ \mu \text{sec.}$

TABLE 1. BO	and heads of the ${}^{3}\Pi(a)-{}^{3}$ AsH and AsD (cm. ⁻¹)	Σ^- transition	of
	AsH	AsD	
³ Π ₀ - ³ Σ-	30,547.0	30,611.1	
³ Π ₁ ⁻³ Σ ⁻	29,959.5	30,049.3	

29,386.0

29,491.8

A group of bands near 3200 Å shows three prominent heads which are attributed to the three sub-bands of the 0-0 band of $a^{3}\Pi(a)$ - $^{3}\Sigma^{-}$ transition

of AsH, by analogy with NH and PH.2,3 The triplet splitting is little changed on deuteration (Table 1), whereas the spacing of the rotational structure is approximately halved. For the configuration $\sigma \pi$, $^{3}\Pi$ the spin-orbit coupling constant, A, is minus one-half the π -orbital coefficient, so that

$$\zeta_{4n}$$
 (As) ~ 1120 cm.⁻¹

The second group of bands consists of a long progression of violet-degraded bands with complex rotational structure. The band spacing decreases by $\sim \sqrt{2}$ on deuteration, and the progressions are assigned to the upper-state bending vibration of the AsH₂ and AsD₂ radicals. The isotope shifts (Table 2) lead to an unambiguous vibrational numbering, the Q-heads being represented to within ± 2 cm.⁻¹ by the equations;

$$AsH_2 \nu = 19,907 \cdot 0 + 858 \cdot 6 v_2 - 3 \cdot 4 v_2^2$$

AsD₂ $v = 19,904 \cdot 9 + 618 \cdot 0 v_2 - 2 \cdot 1 v_2^2$

The maximum intensities in these progressions occur at about $v_2' = 4$ for AsH₂ and $v_2' = 6$ for AsD₂, and use of the Franck-Condon principle indicates a geometry change of $\sim 30^{\circ}$ on excitation. This is very similar to that found for PH2.4 However, the vibrational structure shows no evidence for a marked change in anharmonicity at high values of v_2' as was found in PH₂,⁵ so that the barrier to straightening the AsH₂ molecule must be considerably higher than that in PH2.6 The

- ¹ W. L. Jolly, J. Amer. Chem. Soc., 1961, 83, 335.
 ² R. N. Dixon, Canad. J. Phys., 1959, 37, 1171.
 ³ F. Legay, Canad. J. Phys., 1960, 38, 797.

- ⁴ R. N. Dixon, G. Duxbury, and D. A. Ramsay, to be published.
- ⁵ D. A. Ramsay, Nature, 1956, 178, 374.
- ⁶ R. N. Dixon, Trans. Faraday Soc., 1964, 60, 1363.

rotational structure of the bands near the band heads resembles closely that for PH2,5 but away from the band centres there is evidence for very large doublet splittings, as would be expected from the above value of ζ_{4p} (As).

The spectra are being re-photographed with longer path-lengths in high orders of a 3.4 metre Ebert spectrograph, and rotational analyses are in progress.

TABLE 2. Band heads of AsH_2 and AsD_2 (cm.⁻¹)

	AsH_2		AsD_2	
v_2'	Vvac	ΔG	Vvac	ΔG
0 1	20,762.6	847.4		
2	21,610.0	844·1	21,132.2	606·6
3	22,451.5	835.9	21,738.8	603·2
4 5	$23,287\cdot 4$ $24,116\cdot 1$	828.7	22,342.0 22,942.1	600·1
6	24,935.6	819.5	23,539.2	597·1
7	24,930.0		23,339·2 24,129·4	590·2
8			$24,715 \cdot 2$	585·8 580·6
9			25,295.8	000.0

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